## **Synthesis of Polyacetylenic Acids Isolated from** *Heisteria acuminata*

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**Four linear polyacetylenic compounds were synthesized. Pentadeca-6,8,10-triynoic acid 1 and octadeca-8,10,12-triynoic acid 2 were synthesized by using acetylene coupling reactions. The syntheses of (***Z***)-hexadec-11-en-7,9-diynoic acid 3 and (***Z***)-octadec-12-en-7,9-diynoic acid 4 by using vinylic telluride coupling reactions were accomplished.**

Several examples of polyacetylenic compounds have been isolated in recent years.<sup>1</sup> New linear polyacetilenic compounds **1**, **2**, **3**, and **4** (Scheme 1) were isolated from the



bark of *Heisteria acuminata* by bioassay-guided fractionation.2 These compounds were found to have potent antiinflammatory activity by inhibition of cyclooxygenase (COX) and 5-lipoxygenase (5-LO).3 Polyacetylenes have also been reported previously in the literature as potent inhibitors of the arachidonic acid metabolism.4 Therefore, it may be inferred that compounds  $1-4$  are, at least in part, responsible for the antiinflammatory activity of preparations of *Heisteria acuminata* bark used in folk medicine.5

The unusual structure, the interesting biological activities, and the low availability of these compounds from natural sources encouraged us to synthesize them. In a previous work we have already reported the synthesis of polyacetylenic montiporic acids A and  $B^6$  (Scheme 2). In this Letter we wish to describe an easy route to compounds **<sup>1</sup>**-**4**.

The retrosynthetic analysis of polyacetylenic acids **1** and **2** afforded two basic fragments: alkyldiyne system **B** and



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<sup>(1)</sup> Faulker, D. *J. Nat. Prod. Rep*. **1995**, *12*, 223 and earlier reviews cited therein.

<sup>(2)</sup> Kraus, M. C.; Neszmélyi, A.; Holly, S.; Wiedemann, B.; Nenniiger, A.; Torssell, G. B. K.; Bohlin, L.; Wagner, H*. J. Nat. Prod.* **1998**, *61*, 422.

acetylenic alcohol **A**, which differ only in the number of carbons in the chain (Scheme 3).



The alkyldiynes **B** were synthesized according to Scheme 4. The appropriate terminal acetylene  $(n = 3 \text{ or } 4)$  was



 $a$  (i) *n*-BuLi, THF/I<sub>2</sub>; (ii) pyrrolidine, CuI; (iii) NaOH, xylene, reflux; (iv) *n*-BuLi, THF/I<sub>2</sub>; (v)2 equiv of *n*-BuLi, THF/HMPA; (vi) KNH(CH<sub>2</sub>)<sub>3</sub>NH<sub>2</sub>; (vii) pyrrolidine, CuI; (viii) CrO<sub>3</sub>/H<sub>2</sub>SO<sub>4</sub>, -10 °C.

converted into the 1-iodoacetylene **5** by treatment with *n*-BuLi/iodine<sup>7</sup> in 78% ( $n = 3$ ) and 82% ( $n = 4$ ) yields. Coupling reaction of **5** with **6** in the presence of pyrrolidine/ CuI8 yielded **7** in 95% and 93% yields, respectively, which by treatment with NaOH in xylene under reflux<sup>9</sup> afforded the diynes **8** in 70% and 73% yields. Compounds **8** were then transformed into the corresponding fragment **B** by using again  $n$ -BuLi/I<sub>2</sub> in 85% and 87% yields, respectively.

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The synthesis of fragment **A** (Scheme 4) started with alkylation of the lithium derivative of propargylic alcohol with 1-bromobutane or 1-bromohexane, $^{10}$  yielding the corresponding alcohols **9** in 83% and 89% yields. These compounds were then subjected to prototropic migration of triple bond with KAPA11 to afford the desired terminal acetylenic alcohols (fragment **A**) in 87 and 85% yields. Subsequent coupling reaction of fragments **A** and **B** using CuI and pyrrolidine8 yielded **10** in 90% and 93%. Oxidation with chromium oxide/ $H_2SO_4^{12}$  afforded the desired acids 1 and **2** in 58% and 60% yields, respectively. The overall yield of the sequence was 30% (**1**) and 34% (**2**).

The retrosynthetic analysis of polyacetylenic acids **3** and **4** afforded two fragments: vinylic telluride **C** and 1,3 alkylidyine system **D** (Scheme 5).



The fragment **D** was synthesized following the sequence shown in Scheme 6, by the same procedure used for the synthesis of compound **8** (Scheme 4). The overall yields of fragment **D** were 40% and 43%.

Further, we synthesized fragment **C** (Scheme 6) by hydrotelluration<sup>13</sup> of the appropriate alkyne. The desired

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(15) Compound **1**. <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD)  $\delta$  0.95 (3 H, t,  $J = 7.1$ Hz); 1.43 (2 H, sex,  $J = 7.5$  Hz); 1.54 (2 H, quint,  $J = 7.5$  Hz); 1.58 (2 H, quint, *J* = 7.6 Hz); 1.70 (2 H, quint, *J* = 7.5 Hz); 2.30 (2 H, t, *J* = 7.5 Hz);<br>2.33 (2 H, quint, *J* = 6.9 Hz). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 13.30; 18.92;<br>19 25: 22.00: 24 10: 27 80: 30 20: 60 50: 60 90: 65 83: 66 19.25; 22.00; 24.10; 27.80; 30.20; 60.50; 60.90; 65.83; 66.42; 78.50; 79.83. MS *m*/*z* (%): 230 (55), 128 (100), 91 (50), 41 (32). Compound **2**. 1H NMR  $(400 \text{ MHz}, \text{CD}_3 \text{OD}) \delta$  0.90 (3 H, t,  $J = 7.3 \text{ Hz}$ ); 1.40 (2 H, quint,  $J = 7.5$ Hz); 1.41 (2 H, quint,  $J = 7.5$  Hz); 1.43 (2 H, sex,  $J = 7.3$  Hz); 1.47 (2 H, quint,  $J = 7.5$  Hz); 1.52 (2 H, quint,  $J = 7.3$  Hz); 1.66 (2 H, quint,  $J = 7.3$ Hz); 2.23 (2 H, t,  $J = 7.3$  Hz); 2.28 (2 H, t,  $J = 7.3$  Hz); 2.30 (2 H, t,  $J =$ 7.3 Hz); 13C NMR (100 MHz, CDCl3) *δ* 13.60; 19.25; 19.40; 22.10; 28.15; 28.75; 28.83; 29.80; 30.23; 60.40; 60.50; 65.83; 65.95; 79.20; 79.40; MS *m*/*z* (%): 272 (20), 230 (8), 129 (100), 41 (70). Compound **3**. 1H NMR (400 MHz, CD<sub>3</sub>OD)  $\delta$  0.93 (3 H, t,  $J = 7.2$  Hz); 1.36 (2 H, sex,  $J = 7.3$ Hz); 1.39 (2 H, quint,  $J = 7.5$  Hz); 1.55 (2 H, quint,  $J = 7.3$  Hz); 1.63 (2 H, quint,  $J = 7.6$  Hz); 2.29 (2 H, t,  $J = 7.3$  Hz); 2.35 (2 H, td,  $J = 7.0$ ; 0.7 Hz); 5.48(1 H, dt,  $J = 10.7$ ; 1.1 Hz); 6.07 (1 H, dt,  $J = 10.7$ ; 7.5 Hz). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 13.90; 19.55; 22.30; 24.35; 28.00; 28.43; 30.45; 31.30; 34.05; 65.54; 72.30; 78.24; 84.38; 108.14; 147.81. MS *m*/*z* (%): 246 (8), 146 (30), 129 (50), 117 (100), 91 (68). Compound **4**. 1H NMR (400 MHz, CD<sub>3</sub>OD)  $\delta$  0.93 (3 H, t,  $J = 7.2$  Hz); 1.31 (2 H, quint,  $J = 7.3$  Hz); 1.35 (2 H, quint,  $J = 7.3$  Hz); 1.37 (2 H, sex,  $J = 7.5$  Hz); 1.39 (2 H, quint,  $J = 7.5$  Hz); 1.42 (2 H, quint,  $J = 7.4$  Hz); 1.42 (2 H, quint,  $J = 7.5$ Hz); 1.61 (2 H, quint,  $J = 7.3$  Hz); 2.28 (2 H, t,  $J = 7.4$  Hz); 2.31 (2 H, td,  $J = 7.4$ ; 0.6 Hz); 2.34 (2 H, qd,  $J = 6.9$ ; 1.2 Hz); 5.42 (1 H, dt,  $J =$ 10.9; 1.2 Hz); 6.07 (1 H, dt,  $J = 10.9$ ; 7.6 Hz). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) *δ* 14.02; 19.75; 22.36; 24.87; 28.39; 28.71; 28.88; 29.06; 30.05; 31.00; 65.34; 72.22; 78.29; 84.71; 108.26; 147.95. MS *m*/*z* (%): 271 (10), 245 (6), 187 (30), 164 (27), 117 (100), 91 (65).

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**Scheme 6***<sup>a</sup>*



*<sup>a</sup>* (i) 2 equiv of *n*-BuLi, THF/HMPA; (ii) KNH(CH2)3NH2; (iii) 2 equiv of *n*-BuLi, THF/I2; (iv) pyrrolidine, CuI; (v) NaOH, xylene, reflux; (vi) THF, rt; (vii) EtOH, reflux; (viii) PdCl<sub>2</sub>/CuI, MeOH/Et<sub>3</sub>N; (ix) CrO<sub>3</sub>/H<sub>2</sub>SO<sub>4</sub>, -10 °C.

vinylic tellurides were obtained in a 6:1 mixture of regioisomers, which were separated by column chromatography in 48% yield. These compounds were then coupled to the appropriate alkyldiynes **D** using  $PdCl_2/CuI$  in methanol<sup>14</sup> to afford **11** with the desired *Z* geometry of the double bond. Chromium oxide/ $H_2SO_4^{12}$  (Scheme 4) gave the polyacetylenic acid **3** and **4**. The overall yields of the sequence were 24% (3) and 22% (4). The spectroscopy data<sup>15</sup> (<sup>1</sup>H and <sup>13</sup>C NMR) of compounds  $1-4$  are in agreement with the data reported by Wagner<sup>2</sup> and co-workers.

In summary, we have achieved the total synthesis of four natural products with biological activity and very limited access, using coupling-based methodologies.

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